

HDB-003-001316 Seat No. _____

B. Sc. (Sem. III) (CBCS) Examination November/December – 2017

IC.P-301 : Industrial Chemistry

Faculty Code : 003 Subject Code : 001316

Time: $2\frac{1}{2}$ Hours] [Total Marks: **70**

Instructions:

(1)

- (1) All the questions are compulsory.
- (2) Figures to the right indicate maximum marks.
- (3) Draw labelled diagram wherever necessary.
- (4) Assume suitable data.

1	Answer	the	following	questions.
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- Which group decreases rate of sulfonation reaction?
- (2) Sulphuric acid is used in sulfonation reaction. True/False.
- (3) Describe temperature in recycle apparatus used for hydrogenation of oil____ °C.
- (4) Pd is an oxidizing agent. True/False
- (5) Fluidized bed catalyst for hydrogenation comes under _____.
- (6) In hydrogenation reaction ____ is converted into ____ product.
- (7) Biotic components include _____ (Living organism/Non living organism)
- (8) Which scrubber is suitable when particulates matter is sticky, flammable or corrosive?
- (9) Hydrosphere covers _____ % part of earth surface.
- (10) Los Angeles Smog is also known as _____
- (11) Enlist classification of pollution according to environment.

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	(12)	is extended upto 50 km above from the earth surface.				
	(13)	is natural polymer.				
	(14)	${\rm Fe_2O_3}$ impurity should be avoided for manufacturing of cement.				
	(15)	Highly ductile material possesses percentage elongation.				
	(16)	Bulletproof Glass is type composite material.				
	(17)	Vitreous enamel decreases from surface of ceramics.				
	(18)	Principle constituent of bronze are and				
	(19)	Material of Uniform cross sectional area can be prepared by type molding.				
	(20)	equipment is used to convert cement particles into powder form.				
2	(a)	Answer any three:				
		(1) Explain chemical and physical factors affecting to sulfonation.				
		(2) Define the term halogenation with example.				
		(3) Draw only diagram of Oxygen cycle.				
		(4) Write a short note on Photochemical Smog.				
		(5) Define:				
		(a) Creep				
		(b) Plasticity				
		(6) Explain purpose of alloying with three examples.				
	(b)	Answer any three:				
		(1) Explain various types of oxidation reaction in brief.				
		(2) Draw only diagram of Chloral manufacturing process.				
		(3) Explain sulphur cycle with neat diagram.				
		(4) Draw only diagram of gravitational settling chamber with diagram.				
		(5) Explain weight average molecular and number average molecular weight of polymer with equations.				
		(6) Give types of composite based on their structure. Draw diagram of sandwich composite material				

(c) Answer any two:

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- (1) Describe manufacturing process of naphthalene sulphonic acid from naphthalene.
- (2) Explain continuous sulfonation of benzene.
- (3) Explain cyclone separator with neat diagram.
- (4) Discuss various factors affecting corrosion with examples.
- (5) Explain dry process for manufacturing of cement with diagram.
- **3** (a) Answer any three:

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- (1) Enlist any three oxidizing agents.
- (2) Enlist hydrogenation catalysts.
- (3) Write a note on air pollutants.
- (4) Draw only diagram of packed tower.
- (5) Enlist different types of coatings for protection against corrosion.
- (6) Write Pilling-Bedworth rule with one example.
- (b) Answer any three:

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- (1) Explain flame Ionization detector technique for hydrocarbon in detail.
- (2) Write a short note on Global warming.
- (3) Explain only reactions and parameters of hydrogenation of acids to ester or alcohol.
- (4) Define hydrolysis and enlist various hydrolysis agents.
- (5) Enlist six different properties of material. Give examples of mechanical properties.
- (6) Write types of polymers on the basis of sources with examples and uses.
- (c) Answer any two:

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- (1) Explain source, reaction and effects of SO_x in detail.
- (2) Explain synthesis of methanol in detail with diagram.
- (3) Explain preparation of chlorobenzene from benzene.
- (4) Give a detail account of electrostatic precipitator with schematic diagram.
- (5) Explain injection molding of polymer in detail with diagram, advantages and uses.